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### ADVANCES IN CARBON CAPTURE TECHNOLOGIES FOR MITIGATING CLIMATE CHANGE: A REVIEW

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#### Abstract

*Anthropogenic carbon dioxide emission in the atmosphere cause a major change in the environment such as global warming. The major source of CO<sub>2</sub> emission is coal-fired thermal power plants. Coal combustion produces the flue gas which contains hazardous gases such as CO<sub>2</sub>, SO<sub>2</sub>, NO<sub>2</sub>, etc. CO<sub>2</sub> is the great contributor to the global climate change. Presently, approximately 30 billion tons of CO<sub>2</sub> is generated by us per year. So there is need to develop technologies to mitigate this problem. Carbon capture and sequestration technology used as a mitigation tool to avoid the CO<sub>2</sub> emissions in the atmosphere. To reduce CO<sub>2</sub> emissions three methods are used: Post-Combustion Capture, Pre-Combustion Capture and Oxy-Combustion Capture. These three approaches are at an early stage of development. These technologies remove about 75-90% of CO<sub>2</sub> from coal-fired power plants. Advanced power plants allow many new opportunities for CO<sub>2</sub> capture. Other emerging technologies are also useful in capturing CO<sub>2</sub> in various industries such as absorption, adsorption, membranes and cryogenic separation. In this paper we present the development of the carbon capture and sequestration technologies. The main Moto of these methods is to avoid the CO<sub>2</sub> emissions in the atmosphere.*

**Key words:** CO<sub>2</sub> emission, Carbon dioxide capture, Pre-combustion, Post-combustion, Oxy-combustion.

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#### 1. INTRODUCTION

Increasing CO<sub>2</sub> emission in environment is an issue of great concern today which is leading global warming [1]. The climate variability and changes are one of the evidence for global warming, results to increase in the average temperature rise of the globe [3]. Many climate scientists believe that a major cause of global climate changes is the anthropogenic emission of greenhouse gases (GHGS) into the atmosphere [4]. Green house gases such as water vapour, CO<sub>2</sub>, methane and other atmospheric gases absorb outgoing infrared radiation which causing an increase of earth's temperature, this phenomenon called as Greenhouse effect[1]. Being a member of GHGs, CO<sub>2</sub> is the major contributor for global warming and it has the greatest antagonist which accounts approximately 55% of the observed global warming. About 64% of the only CO<sub>2</sub> is responsible for the enhanced Greenhouse effect [1]. The emission of greenhouse gas is increased when comparing with the initial revolution of the industries [3]. About 60% of the CO<sub>2</sub> is emitted by industrial power system, due to the usage of fossil fuels [5]. Coal is mainly used in the power sector and it consumes 70% of the Indian economy. Looking into current scenario it is necessary to minimize the global warming effect due to emission of CO<sub>2</sub> [3]. Currently, approximately 30 billion tons of CO<sub>2</sub> each year is generated by us [18]. And supply of about 80% of the world energy until 2040 is depending on the fossil fuels. [11]

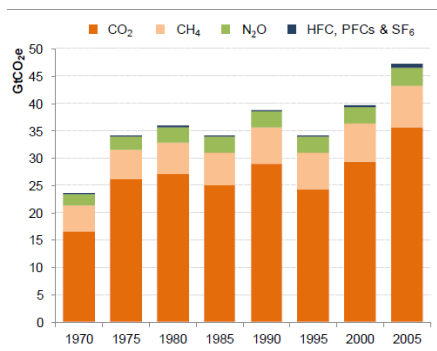
In 1990, nearly 80% of global warming potential (GWP) weighted CO<sub>2</sub> emission from fossil fuel combustion. From 1990 to 1999 emission of CO<sub>2</sub> from this source was increased about 13% [10]. During November 1991, the International

Energy Agency (IEA) Greenhouse Gas (GHGS) programme launched and it has 17 member countries and 7 industrial sponsors [8]. This International co-operation focus to identify and evaluate technologies for reducing emission of GHGs which arising from the usage of fossil fuel [8]. Its Primary technical focus is carbon capture and sequestration [8]. The Energy Information Administration (EIA) within the U.S. Department of Energy (DOE) appraise that consumption of fossil fuel such as coal, petroleum and natural gas will increases by 27% over the next 20 years, thus increasing U.S. CO<sub>2</sub> emission from the current 6000 million tonnes per year to 8000 million tonnes per year by 2030. Especially the EIA estimates that the combined CO<sub>2</sub> emission from China and India in 2030 from coal use will be three times that of the United States. According to advance declaration of Intergovernmental panel on climate change (IPCC), by the year 2100, the atmosphere may contain up to 570 ppmv of CO<sub>2</sub> causing a rise of mean global temperature of around 1.9°C and an increase in mean sea level of 3.8m[1]. It has becomes an important research issue of global approach as more international attention is focused on global warming [1]. It has been predicted that the global average rise temperature will be 0.3-0.7 °C between 2016 and 2035 [16]. In order to reduce the increase in temperature of globe which causing global warming under a threshold of 2°C, carbon capture and sequestration technologies may be implemented with less risk [20].

One approach that keep promise for reducing GHG emission is Carbon Capture and Sequestration (CCS)[4]. Carbon dioxide capture and sequestration is the process of separation

of CO<sub>2</sub> from the emitting sources of CO<sub>2</sub> and that captured CO<sub>2</sub> is then stored in the carbon sinks [2]. CO<sub>2</sub> will be stored in underground sedimentary basins, saline aquifers and coal reservoirs [3]. The main target of the CCS technology is to reduce the CO<sub>2</sub> concentration in the atmosphere [3]. This approach brings these sequestered CO<sub>2</sub> storage for thousands of years [4]. Advanced coal power plants allow many new opportunities for CO<sub>2</sub> capture. For example to integrate CO<sub>2</sub> capture with an Integrated Gasification Combined Cycle (IGCC) [21]. Current capture technologies can remove about 75 to 90% of the CO<sub>2</sub> emitted from the combustion of fossil fuels. By reducing the cost of capture and sequestration technologies motivates the both arrangements of capture and sequestration technologies [19]. The advanced plant designed reduces the CCS energy requirement and will also the overall environmental impact and cost [14]

National Energy Technology Laboratory (NETL) organized the Department of Energy (DOE) carbon sequestration program which follows the five technological paths for reducing the GHG emissions. They are CO<sub>2</sub> separation & capture, carbon sequestration (storage), monitoring, mitigation and verification of stored CO<sub>2</sub>, control of non-CO<sub>2</sub> GHGs and breakthrough concepts related to CCS (Figuroa et al. 2007) DOE's targeting the mandatory technology ready for the large scale field testing. It should become necessary to force required limits on CO<sub>2</sub> emission. Three technological pathway follows for CO<sub>2</sub> capture from the coal used power plants are post-combustion capture, pre-combustion capture and oxy-combustion. These three pathways are under the continue research and development to decline the GHG emissions and to significantly improve performance with great cost reduction [4].



**Fig-1: GHG Emission (By Gases) 1970-2005[5]**

There are various existing and emerging technologies available for CO<sub>2</sub> capture such as adsorption, absorption, cryogenic separation, gas separation membrane etc. For improving overall reduction of CO<sub>2</sub> (in the absorption process) there is need to consider the advancement in the absorption process such as; development of new solvent, development of suitable corrosion inhibitors and modification of the processes. If we know the composition of CO<sub>2</sub> flue gas streams, quantity of CO<sub>2</sub> capture and uses of captured CO<sub>2</sub>, then we can recommend a suitable technology for CO<sub>2</sub> capture. (Mondal et al. 2012). All these existing technologies need to improve the performance, reduction of cost and energy for CO<sub>2</sub> separation [1].

This paper reviews the existing and emerging technologies for the separation and capture of CO<sub>2</sub> from the power plants. The

main goal of the carbon capture and sequestration technologies is to reduce the amount of CO<sub>2</sub> emission in the environment.

### 1.1 Motivation For CO<sub>2</sub> Capture And Sequestration

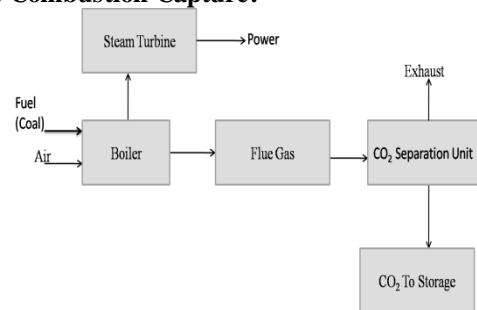
Coal fired power plants are the major source of CO<sub>2</sub> emissions which causing a major change in the environment such as global warming. There is need to reduce carbon emissions from the fossil fuels. To eliminating the CO<sub>2</sub> emission from the fossil fuel power plants, cost effective carbon capture and sequestration technology used as a key need for dealing with CO<sub>2</sub> impact on global climate change. These technologies capture the CO<sub>2</sub> from fossil fuel power plants and stored the captured CO<sub>2</sub> in the geological formations or oceans [19]. Based on this motivation, we studied the advanced carbon capture and sequestration technologies.

### 1.2 Objective

- Reducing the CO<sub>2</sub> emissions by applying CCS to power plants.
- Utilization of CO<sub>2</sub>.
- To reduce future retrofit costs, new plant designs could take future CCS application into account.

## 2. CO<sub>2</sub> CAPTURE TECHNOLOGIES

### 2.1 Post-Combustion Capture:



**Fig-2: Post-Combustion Capture**

Post-combustion capture method is used to remove the CO<sub>2</sub> from the flue gases. Air used in the power plants consist almost four-fifth of Nitrogen for the combustion (Figuroa et al. 2007) and generate a flue gas at atmospheric pressure and typically having the CO<sub>2</sub> concentration less than 15%. In the power generation system coal is burnt in the boiler with the help of air, for the generation of the steam. Steam drives the turbine and it generates the electricity. [3]. The exhaust contain CO<sub>2</sub> and N<sub>2</sub> flue gases. The greatest challenge in this method is the separation of the low concentration of CO<sub>2</sub> in power plant flue gas (13-15% for coal fired power plant) out of the high concentration of N<sub>2</sub> gas (73-77%). [1,6]. Post combustion carbon capture having the high potential to reduce the GHG emissions in the power sector. Because it can be retrofitted to existing plants that generates two-third of the CO<sub>2</sub> emissions. [1]. Post-Combustion capture technology is best as compared to the other technologies in the coal-fired power plants. Because it has a higher thermal efficiency for conversion to electricity than the pre-combustion capture technology [3,1]. The post combustion capture technology can also be used in other industries other than power plants like cement, petrochemicals and oil refining [7].

### Advantages:

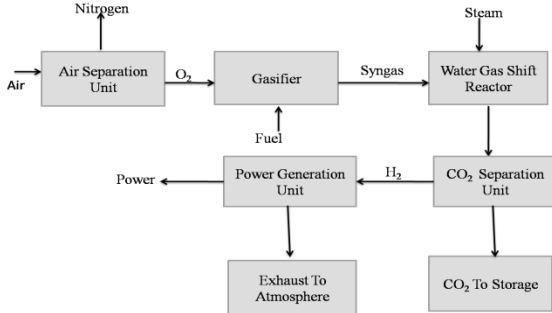
- Applicable to existing power plants for continued operations.
- Retrofit technology option.
- Simple process.

- More process available.

**Disadvantages:**

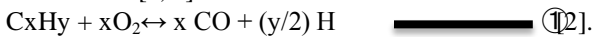
- Plot space requirements are significant for post combustion capture.
- Extra costs may be required to fit post-combustion capture at plant location.
- Low CO<sub>2</sub> level.

**2.2 Pre-Combustion Capture:**



**Fig-3: Pre-Combustion Capture**

In the pre combustion CO<sub>2</sub> capture, the CO<sub>2</sub> is captured from some process stream, before the fuel is burned (Figuroa et al. 2007). Primarily Oxygen and Nitrogen is separated in the air separation unit. Then the fuel or coal is mixed with oxygen in the gasifier. This mixture forms the synthesis gas at the high temperature as seen in equation (1). It is also called as ‘Syngas’. Syngas is the mixture of CO and H<sub>2</sub>. This process is called as gasification [3]. Then steam is added in the water-gas reactor where CO is converted into CO<sub>2</sub>, as seen in equation (2). In the water gas shift reactor, the addition of steam and decrease in temperature cause the conversion of CO into CO<sub>2</sub> and then it increase the yield of H<sub>2</sub> [1]. In this mixture, 15-50% range of CO<sub>2</sub> is present. After the pre combustion method, CO<sub>2</sub> is separated, transferred, and then sequestered or stored. [3]. This occurs at a high pressure. (1000 times higher than flue gas in post-combustion) [6]. Pure H<sub>2</sub> is used as a fuel in a gas-based the power plant to produce electricity [1,2]. Hydrogen is also used in the power fuel cells for rising the overall plant efficiency. In future, H<sub>2</sub> would be used as transportation fuel (Mondal et al, 2012). Pre-combustion capture is the more efficient technique for the removal of CO<sub>2</sub> at low pressure from the syngas due to the presence of CO<sub>2</sub> at higher concentration in syngas. So the CO<sub>2</sub> capture is less expensive for pre-combustion as compared to the post-combustion [1, 3]



**Advantage:**

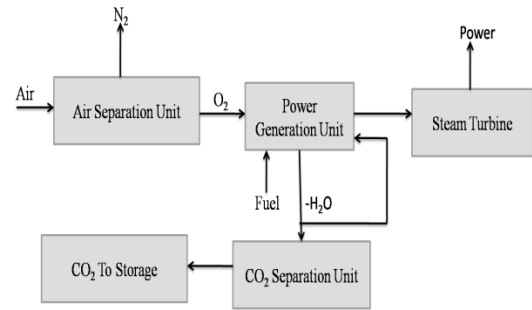
- 90 % CO<sub>2</sub> is captured.
- High CO<sub>2</sub> level.
- Easy separation of CO<sub>2</sub>.
- H<sub>2</sub> used as input to produce electricity.
- Less water is used as compared to post combustion capture.

**Disadvantage:**

- Applicable mainly to new plants.
- Complicated process.
- Equipment cost is high.

- Applications of gasification are common to pre-combustion capture.

**2.3 Oxy-fuel Combustion Capture:**



**Fig-4: Oxy-fuel Combustion Capture**

Combustion of fuel with oxygen is called as oxy-combustion [7]. About 3-15% of conc. CO<sub>2</sub> is present in the fuel gases. For the removal of concentrated CO<sub>2</sub> from the flue gases oxy-fuel combustion method is used [2]. Air separation unit is used to supply the high purity oxygen to the coal fired power plants [4]. The fuel is burned with the pure oxygen (> 95 %) and then this pure O<sub>2</sub> mixed with the recycled flue gas (RFG) in the boiler to maintain the combustion conditions similar to an air fired configuration [1, 4]. This is necessary because currently available materials of construction cannot withstand the high temperatures resulting from coal combustion in pure oxygen (Figuroa et al. 2007). The advantage of this method is that, after the combustion, the flue gas consist CO<sub>2</sub> and water vapour. The water is easily removed by the condensation and remaining CO<sub>2</sub> can be extracted easily at low cost [1]. In an oxy-fuel combustion capture technology, the carbon capture cost is very less as compared to other technologies. But air separation and recycling of flue gas significantly increases the economic cost. Currently oxygen production techniques are very expensive, in order to uses this oxy-fuel combustion technology a wide range of advancement must be required [6].

**Advantage:**

- Very high CO<sub>2</sub> conc. in flue gas.
- Retrofit technology option.
- 98% CO<sub>2</sub> captured.
- Cost of separation is low.
- Oxy-combustion with CO<sub>2</sub> capture should be at least competitive with pre- and post-combustion CO<sub>2</sub> capture.

**Disadvantage:**

- Pure form of oxygen is required in power plants.
- Recycling of flue gases increases the cost process.
- Material of construction to withstand at high temp. is needed.

**3. EXISTING TECHNOLOGIES:**

**3.1 Absorption Process:**

The mostly used technology for CO<sub>2</sub> capture from flue gas is absorption by using the aqueous amine solution like diethanolamine (DEA), monoethanolamine (MEA) and methyl diethanolamine (MDEA). About 98% of CO<sub>2</sub> is recovered by using MEA [13]. This technology is used in the post-combustion capture technology. The major limitation of this technology is that, the impurities present in the flue gases degrade the solvent and affect the performance of solvent. To overcome these limitations, aqueous ammonia as a solvent is used for the CO<sub>2</sub> separation. Ammonia also absorbs the other

impurities like NO and SO<sub>x</sub> from the gas stream. The chilled ammonia causes the fouling of the heat exchanger by the deposition of the ammonium bicarbonate [11]. Aqueous and chilled ammonia processes have the great potential of improve energy efficiency as compared to amine based system [4].

### 3.2 Adsorption Process:

Removal of the one or more components of a mixture by using a solid surface called as adsorption process [1]. A packed column is used in the adsorption process to capture CO<sub>2</sub>. Spherical adsorbent is filled in the column. When the CO<sub>2</sub> containing stream is fed to the column, CO<sub>2</sub> is attracted towards the adsorbent and sticks on the surface of adsorbent. At equilibrium, to get the CO<sub>2</sub> in the pure form desorption takes place and adsorbent are regenerated for the further process [1]. Gas is adsorbed on the solid surface of the adsorbents by a van-der Waals force. The most effective adsorbents are zeolites, activated carbon, silica gel and aluminium oxide [15].

Zeolites are mostly used as solid sorbents in CO<sub>2</sub> capture application. CO<sub>2</sub> adsorbed on the zeolites by different procedures [11]. Some factors like adsorbent pore size pore volume, surface area and affinity of the adsorbed gas for the adsorbent are controlled the adsorption capacities and kinetics [12].

### 3.3 Membrane Technology:

Membranes are the semi permeable barriers which separate the CO<sub>2</sub> and other impurities from the flue gas. The substances are separated by the various mechanisms such as solution, adsorption, diffusion, molecular sieve and ionic transport [1]. If flue gas is pass through the bundle of membrane tubes, when an amine solution flowed into the shell side of the bundle. Then CO<sub>2</sub> would pass through the membrane and absorbed in the amine while impurities blocked from the amine (Figuroa et al. 2007). Amine leaves the membrane bundle and would be regenerate hurriedly before it recycled [1, 4].

Two types of membrane process can be used for separating CO<sub>2</sub> from flue gas:

- Gas separation membrane
- Gas absorption membrane.

#### 3.3.1 Gas Separation Membrane:

It operates on the principle of diffusion [17]. The pores of membrane permit the preferential permeation of mixture constituents. These pores of the membrane permeate the one component through the membrane faster than the other component. The principal design and operational parameter of membrane are selectivity and permeability [1, 17]. This technology is not favourable to separate CO<sub>2</sub> from flue gas because of the large quantity of the flue gas and compression energy requirements [15]. Gas separation membrane consisting ceramic, polymeric and combination of both materials or mixed matrix membranes. [1]

#### 3.3.2 Gas absorption Membrane:

Gas absorption membrane consists of micro porous solid membrane. These membranes are used for the contacting of the gas and aqueous absorbent [1, 17]. CO<sub>2</sub> is diffuses through the membrane and absorbed by the mono-ethanolamine (MEA). This process removes high amount of CO<sub>2</sub> than the

gas separation membrane because of the high driving force at any moment [1, 17].

**3.4 Cryogenic Separation:** Cryogenic separation process is mostly used for the separation of CO<sub>2</sub> from the stream which contains the high CO<sub>2</sub> concentrations (>50%) [1]. This method uses the different boiling point of a various gas species to separate them from gas mixture [15]. This process includes compression and cooling of the gas mixture in various stages to induce phase changes in CO<sub>2</sub> in flue gases [1, 17]. CO<sub>2</sub> can be arises as a solid or liquid phase together, depending on the operation conditions [1]. The major advantage of this method is that no chemical absorbent is required and process can be operating at atmospheric pressure. This process gives the high recovery of CO<sub>2</sub>. The cost of this process may not be higher than amine absorption process [17].

### 4. CONCLUSION

In order to reduce the growth of CO<sub>2</sub> emissions, carbon can be converted into a supportable form or captured as a gas and stored in the geological formations or oceans. Post combustion technology is the most suitable technology as compared to the pre-combustion and oxy-combustion technology in coal fired power plants. CO<sub>2</sub> is separated from the nitrogen at low pressure and it has the high potential to reduce the GHG emissions from the power sectors. This technology can be retrofitted to the existing plants. Other emerging technologies are also helpful for separation of CO<sub>2</sub>. All these existing technologies have their own advantages and limitations. But their accuracy, strength and removal efficiency are the main challenges. There is need to improve the performance and reduce the cost for the CO<sub>2</sub> separation.

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